

Gas-fired chemical looping combustion with supercritical CO₂ cycle

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Abstract

Oxy-fuel combustion is currently gathering attention as one of the promising options for capturing CO₂ efficiently, when applied to power plants, for subsequent carbon sequestration. However, this option requires a large quantity of high-purity oxygen that is usually produced in an energy-intensive air separation unit (ASU). Chemical looping combustion (CLC) is a technology with the potential of reducing the costs and energy penalties associated with current state-of-the-art cryogenic ASUs. In this work, the techno-economic performance of a natural gas-fired oxy-combustion cycle with cryogenic ASU is compared with that based on CLC. Two natural gas-fired cycles are considered: (i) staged oxy-fuel natural gas combined cycle as a reference; and (ii) gas-fired CLC with supercritical CO₂ cycle. The process models were developed in Aspen Plus® in order to evaluate the thermodynamic performance of the proposed system and to benchmark it against the reference cycle. The results show that the net efficiency of the proposed cycle, including CO₂ compression, is more than 51%, which is comparable to that of a conventional natural gas combined cycle with CO₂ capture and 2.7% points higher than that of the reference cycle. Moreover, the economic evaluation indicates that a reduction in levelised cost of electricity from £38.3/MWh to £36.1/MWh can be achieved by replacement of the ASU-based oxy-fuel system with CLC. Hence, gas-fired CLC with a supercritical CO₂ cycle has high potential for commercialisation.

Key words: Carbon capture, oxy-fuel turbine, cryogenic ASU, chemical looping combustion, natural gas combined cycle, oxygen production

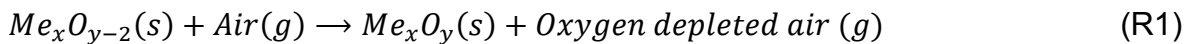
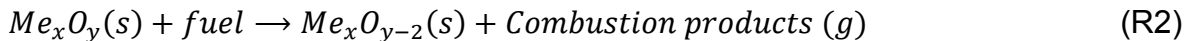
1. Introduction

Carbon capture and storage (CCS) is increasingly regarded as an effective option to limit greenhouse gas emissions, while still maintaining the ability of utilising fossil fuels for power generation. The potential strategies proposed for CCS include pre-combustion, post-combustion and oxy-combustion [1].

Oxy-combustion has become one of the most promising options applicable to power generation systems [2]. This technology involves the process of fuel combustion in a high-purity O₂ environment that results in the flue gas containing mostly pure CO₂ after water vapour has been separated by condensation. Importantly, fuel combustion under such conditions would result in intolerably high temperatures in the combustors, reaching more than 2000°C. Therefore, some fraction of the exhaust gas is recycled back into the combustor to moderate the flame temperature.

It needs to be stressed, however, that the oxy-combustion process requires high O₂ production capacity, as for a 500 MW_e power plant the O₂ requirement is almost 10,000 t/d [3]. Currently, the cryogenic air separation unit (ASU) is the only commercially applied and mature technology that is capable of producing such large quantities of O₂ at high purity [4], but it is a complex and energy-intensive technology, the power requirement for which is about 200 kWh/tO₂ at 95%_{vol} O₂ purity [5].

In chemical looping combustion (CLC), O₂ is separated from the air and then transported to the fuel via an oxygen carrier (OC). In its preferred embodiment, CLC consists of two interconnected fluidised bed reactors: an air reactor and a fuel reactor. O₂ separated from the air in the air reactor is transferred by OCs to the fuel reactor [6]. The chemical reactions in the air and fuel reactors, respectively, are as follows:



The reduced OC is then recycled to the air reactor. The gas stream leaving the fuel reactor comprises mainly CO₂ and water vapour. The benefit of using CLC is that the combustion products (CO₂ and water vapour) are inherently separated from other components such as N₂ and Ar. Hence, unlike in the case of post-combustion capture (PCC), no additional energy is required for CO₂ separation [7].

CLC is a promising technology [8] which has been developed for combustion of gaseous, liquid and solid fuels and is undergoing significant scale-up at present [9]. The largest (120 kW) gas-based CLC system is located at the Vienna University of Technology [10]. Furthermore, a number of OCs have been thoroughly studied. Källén et al. [11,12] tested calcium manganite in a 10 kW pilot-scale CLC unit and iron/manganese/silicon-based OCs in a 300 W lab-scale CLC unit. All performed well with respect to gas conversion, achieving high O₂ transport capacity. Furthermore, regarding the performance of different CLC configurations, Brandvoll and Bolland [13] revealed that a net efficiency of 54% was obtainable with integration of CLC to a humid air turbine cycle. Naqvi and Bolland [14] showed that using CLC in the conventional combined cycle can result in a net efficiency of 53%, including CO₂ compression. Chen et al. [15] analysed the possibilities and benefits of integration of CLC with supercritical CO₂ for combined heat and power cogeneration. Their proposed cycle reached a net power efficiency of 41.3% with a heating efficiency of 40.4%. However, the economic feasibility of such concept has not yet been proven. Spallina et al. [16] performed a thermodynamic analysis on the integration of solid oxide fuel cells (SOFCs) with CLC in natural gas-fired power plants. The integrated plant showed a net efficiency in the range of 63-70%. Hamers et al. [17] analysed a two-stage CLC integrated with a coal gasification plant. They revealed that a two-stage CLC can achieve a net efficiency of about 40% and results in significantly lower reactor cost compared to a single-stage CLC. Olaleye and Wang [18] performed an economic analysis of a CLC unit integrated with a humid air turbine that was characterised with an efficiency of 57.1%. Their study has shown that for a 50 MW_{th} plant, the capital cost and net present value are £52M and £104M, respectively, assuming a cost of electricity of £77.5/MWh. However, to support commercialisation of the CLC technology, its techno-economic feasibility needs to be further evaluated, especially considering advanced power cycles, such as the supercritical CO₂ cycle (sCO₂) that is considered as a suitable replacement for conventional steam cycles [19].

This work evaluates the techno-economic feasibility of integrating CLC with the sCO₂ cycle to achieve high efficiency with low CO₂ emissions and affordable electricity cost. A novel gas-fired chemical looping combustion process with supercritical CO₂ cycle (CLC-sCO₂) is proposed and its performance is benchmarked against staged oxy-fuel natural gas combined cycle (SOF-NGCC). The process model of the CLC-sCO₂ is

developed in Aspen Plus[®]. Parametric studies are performed to determine the optimum thermodynamic performance, characterised with the highest net efficiency. Finally, the economic feasibility of the considered cycles is evaluated and compared in terms of capital cost and levelised cost of electricity. Therefore, this work demonstrates the advantage of using CLC instead of an ASU-based oxy-fuel combustor for high-efficiency low-CO₂-emission power generation. It also proves the economic feasibility of linking CLC with the sCO₂ cycle that is characterised by a higher efficiency and smaller size compared to other cycles, such as the conventional steam cycle. Such information on process design, operation, and techno-economic feasibility will support further development of CLC for power generation.

2. Process description and simulation

Figure 1 presents a schematic of the SOF-NGCC. The entire quantity of high-purity O₂, required to ensure complete combustion, enters into the first combustion stage, whereas natural gas enters into each of three combustion stages with nearly equal feed rates. The combustion products of the first stage, along with unreacted O₂, act as a diluent for the second-stage combustion. Additionally, to maintain the desired combustion temperature of the topping cycle, excess heat is extracted from the first stage to pre-heat the CO₂ stream in the sCO₂ cycle primarily by radiation as described by Gopan et al. [20,21]. Absence of the exhaust gas recycle (EGR) results in a high combustion temperature and heat transfer. This may result in a high surface temperature that may exceed allowable operating limits. This challenge is addressed by staging the delivery of the fuel and controlling the characteristics of the flame [22]. This process continues until O₂ is almost entirely (except for 5% excess O₂) consumed at the last combustion stage. The exhaust gas, which is mainly composed of water vapour and CO₂, is expanded in the turbine to generate power. The flue gas then passes through a heat exchanger and transfers heat to the O₂, fuel and CO₂ streams in the sCO₂ cycle. The water vapour is easily separated from the cold flue gas and the remaining CO₂ is sent to the carbon purification unit (CPU) to be conditioned for storage. In the bottoming cycle, the sCO₂ stream is first pressurised to 300 bar and then preheated in a high-temperature recuperator (HTR) and low-temperature recuperator (LTR), as well as multi-stage combustor, before entering the high-

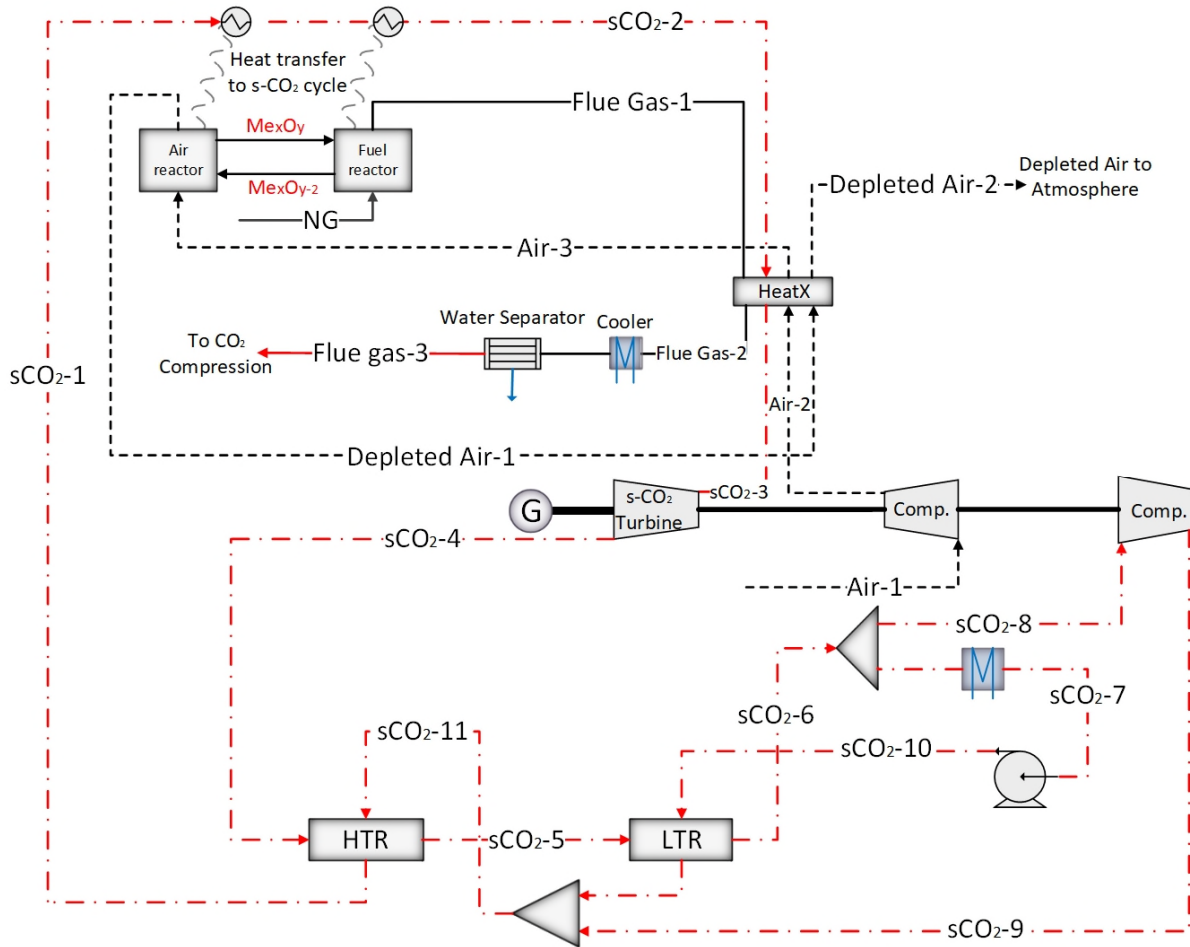


Figure 2. Schematic of CLC-sCO₂ (sCO₂ gains heat, Depleted Air-1 loses heat, Flue gas-1 loses heat, Air-2 gains heat, sCO₂-4 loses heat, sCO₂-5 loses heat)

2.1. Model development

The process model for CLC-sCO₂ has been developed in Aspen Plus®. The package used for the thermodynamic property estimation is the Peng Robinson equation of state which is suitable for hydrocarbons and light gases, such as CO₂ and H₂ [24]. All components are defined as conventional except for OCs which are solid. The reactors are modelled as Gibbs reactors (*RGibbs*), which assume chemical and phase equilibrium based on Gibbs energy minimisation [25]. Importantly, it is assumed that heat loss is negligible. All heat exchangers are modelled using the *MHeatX* block and designed based on the assumption that the minimum temperature approach of the heat exchangers is 5°C, similarly to previous studies on the sCO₂ cycle [26,27]. Similarly to the study by Hanak and Manovic [19] and Le Moullec [28], it is assumed

that high-pressure CO₂ in the sCO₂ cycle is heated in the CLC reactors. All turbines are modelled as individual turbine sections using the *Compr* block. Cu has been selected as OC for this study due to its high reaction rate, high O₂ transport capacity and its complementary combustion thermodynamics that completely convert fuel to CO₂ and water vapour [29,30]. It is assumed that the agglomeration rate of Cu is negligible. Moreover, due to the high oxidation conversion rate of Cu [31] and its high combustion temperature, it is assumed that the complete oxidation of Cu occurs [32]. The properties of the natural gas, key assumptions used in modelling, the turbomachinery and initial simulation parameters for the CLC-sCO₂ are summarised in Table 1.

Importantly, the SOF-NGCC, which has been thermodynamically analysed by Khallaghi et al. [23], is selected as a reference cycle in this study. To have a fair comparison between CLC-sCO₂ and the reference cycle, the SOF-NGCC simulation is adapted to the new assumptions mentioned in Table 1.

Table 1. Main assumptions and turbomachinery specification for simulation

Parameter	Value
Natural gas composition and conditions [23]	
Methane (% _{vol})	89
Ethane (% _{vol})	7
Propane (% _{vol})	1
Butane (% _{vol})	0.1
Pentane (% _{vol})	0.01
CO ₂ (% _{vol})	2
N ₂ (% _{vol})	0.89
Lower Heating Value (LHV) (MJ/kg)	46.50
Temperature (°C)	15
Pressure (bar)	1.25
Turbomachinery specification	
Isentropic efficiency of pump (%) [33]	90
Isentropic efficiency of turbine (%) [33]	93
Isentropic efficiency of compressor (%) [33]	89
Mechanical efficiency of compressors and pump (%) [34]	99.6
Electrical efficiency of generator (%) [35]	98.5
Initial CLC-sCO₂ operating parameters	
Oxygen carrier type	Cu/CuO
Oxygen carrier mass flow rate (kg/s)	350*
Combustors (oxidation and reduction) pressure (bar)	1.25
Combustors and reactor pressure drop (mbar)	150
Pressure drop in heat exchangers (%)	1
Oxidation reactor temperature (°C)	995
Reduction reactor temperature (°C)	900
Turbine backpressure (bar)	35
sCO ₂ turbine inlet temperature (°C)	700
sCO ₂ turbine inlet pressure (bar)	300
sCO ₂ turbine backpressure (bar)	75
Recompression split fraction (-)	0.3

*Assumption is made based on fully oxidised oxygen carrier flowing into the fuel reactor.

2.2. Model validation

The considered process consists of two parts: (i) the power generation, which is based on the sCO₂ cycle, and, (ii) CLC that converts the chemical energy of natural gas into heat for the sCO₂ cycle. Importantly, the prediction of the CLC model is comparable to that presented in the work by Mantripragada and Rubin [10]. The sCO₂ cycle has been developed based on the recompression sCO₂ cycle from Moisseytsev and Sienicki [36] and has been validated by Hanak and Manovic [19]. As this study considers a pump in place of the main compressor, the data from Moisseytsev and Sienicki [36] are used as a benchmark for the prediction of the sCO₂ cycle model used in this study (Table 2).

Table 2. Benchmark of the sCO₂ cycle stream data with Moiseyev and Sienicki [36].

Stream	Temperature		Pressure	
	Literature	Model	Literature	Model
sCO ₂ -7	31.3	31	74.0	75
sCO ₂ -10	84.4	52	200.0	200
sCO ₂ -11	171.8	178	199.6	198
sCO ₂ -1	323.3	317	199.1	196
sCO ₂ -3	471.8	472	198.4	198
sCO ₂ -4	362.3	364	77.3	77
sCO ₂ -5	190.7	190	76.9	76
sCO ₂ -6	90.2	85	76.3	75

Using the maximum cycle pressure of 200 bar for the stream entering LTR and sCO₂ turbine inlet temperature of 472°C as in Moiseyev and Sienicki [36], the temperatures and pressures of the streams in the sCO₂ cycle are compared with the data reported in that study. The main reasons for such a deviation are different turbomachinery efficiencies and pressure drops considered in this study compared to those in Moiseyev and Sienicki [36]. Moreover, a pump and a compressor are used in this study instead of two compressors for the compression stage. This resulted in a lower temperature of the CO₂ stream entering the LTR. Nevertheless, the model prediction was found to be in good agreement with the literature data.

3. Techno-economic performance indicators

To evaluate the thermodynamic performance of the CLC-sCO₂, the thermal efficiency of the system is defined in Eq. (1) as the ratio of the net power output ($\dot{W}_{net}$), which is calculated as the gross power output less the system's parasitic load, and the chemical energy input to the system, that is defined as the product of the fuel consumption rate ($\dot{m}_{fuel}$) and its lower heating value (LHV). The gross power output of the CLC-sCO₂ is the electric power output of the generator linked with the sCO₂ turbine through the mechanical shaft. Importantly, the parasitic load is the sum of all parasitic loads of the entire cycle associated with its compressors and pumps.

$$\eta_{net} = \frac{\dot{W}_{net}}{\dot{m}_{fuel} \cdot LHV} \quad (1)$$

The economic performance of the considered cases is evaluated using the levelised cost of electricity (LCOE), defined in Eq. (2). This equation is based on the assumption that the capital cost of the proposed process is completely covered by the revenue from electricity sales. Therefore, the LCOE indicates the minimum electricity price required for the system to become economically feasible. This approach correlates the net power output, net thermal efficiency, and capacity factor (CF) with economic performance indicators, such as total capital requirement (TCR), variable (VOM) and fixed (FOM) operating and maintenance costs, specific fuel cost (SFC) and the fixed charge factor (FCF). The main assumptions for the economic analysis are described in Table 3.

$$LCOE = \frac{TCR \times FCF + FOM}{8760 \times \dot{W}_{net} \times CF} + \frac{SFC}{\eta_{net}} + VOM \quad (2)$$

Table 3. Assumptions for the economic analysis [34]

Parameter	Value
Variable operating cost as a fraction of total capital cost (%)	2.0
Fixed operating cost as a fraction of total capital cost (%)	1.0
Natural gas price (£/GJ)	3.0
Plant lifetime (years)	25
Project interest rate (%)	8.75
Capacity factor (%)	80

Importantly, FCF, which divides the total capital cost into uniform annual amounts over the project lifetime, is calculated using Eq. (3) considering the project interest rate (r) and project lifetime (T). Finally, the annual net electricity generation, FCF along with the fuel costs, and operating and maintenance costs, are assumed to be constant for the project lifetime.

$$FCF = \frac{r(1+r)T}{(1+r)^T - 1} \quad (3)$$

The capital costs for the considered cases are estimated using the bottom-up approach, considering the individual capital costs of the key equipment. The capital costs for combustors in the SOF-NGCC and reactors in the CLC-sCO₂ are estimated

using the component scaling factor, as shown in Eq. (4). In the latter case, both reactors are assumed to be fluidised bed reactors. Having assumed that the residence times for the air reactor and fuel reactor are 5 and 60 seconds [37], respectively, the volume for each reactor is calculated using the procedure described by Lyngfelt et al. [38]. Then, the weight for each reactor is calculated as presented in Peters and Timmerhaus [39]. Thereafter, by considering reactors described by NRE [40] as a reference and using the scaling factor of 0.67, the capital cost of both CLC reactors is estimated. On the other hand, the reference for capital cost of combustors used in the SOF-NGCC is selected from López et al. [41]. Considering the total outlet mass flow rate as a scaling parameter and the scaling factor of 0.6, the cost for the combustor in the SOF-NGCC is calculated.

cost of equipment A

$$\begin{aligned}
 &= (\text{cost of equipment } A_{ref}) \\
 &\times \left(\frac{\text{capacity of } A}{\text{capacity of } A_{ref}} \right)^{\text{scaling factor}}
 \end{aligned} \tag{4}$$

The capital cost of the SOF-NGCC topping cycle is determined based on the capital cost correlations for specific pieces of equipment, using the bottom-up approach. These correlations were taken from the literature and are gathered in Table 4.

Table 4. Capital cost estimation for SOF-NGCC topping cycle

Equipment [scaling parameter]	Correlation
Fuel compressor [Brake power requirement, $\dot{W}_{FC,BRK}$ (kW)] [42,43]	$C_C = 91,562 \left(\frac{\dot{W}_{FC,BRK}}{445} \right)^{0.67}$
Air/O ₂ compressor [Brake power requirement, $\dot{W}_{AC,BRK}$ (kW)] [42,43]	$C_C = 91,562 \left(\frac{\dot{W}_{AC,BRK}}{445} \right)^{0.67}$
Air separation unit [O ₂ production rate, $\dot{m}_{O_2}$ (kg/s)] [44]	$C_{ASU} = 2.926e^7 \left(\frac{\dot{m}_{O_2}}{28.9} \right)^{0.7}$
Oxygen pressure booster pump [Break power output $\dot{W}_{OP,BRK}$ (kW), Isentropic efficiency η_{OP} (-)] [45]	$C_{OP} = 623.22 (\dot{W}_{OP,BRK})^{0.95} \left(1 + \frac{0.2}{1 - \eta_{OP}} \right)$
Turbine [Break power output, $\dot{W}_{T,BRK}$ (kW)] [46]	$C_T = 3744.3 (\dot{W}_{T,BRK})^{0.7} - 61.3 (\dot{W}_{T,BRK})^{0.95}$
Generator [Break power output, $\dot{W}_{T,BRK}$ (kW)] [46]	$C_{Gen} = 26.18 (\dot{W}_{T,BRK})^{0.95}$

The capital costs associated with the SOF-NGCC bottoming cycle and the power cycle of the CLC-sCO₂, both of which are sCO₂ cycles, are determined from the capital cost correlations for each specific piece of equipment as outlined below. To estimate the costs of the sCO₂ turbine and compressor, the methodology presented by Benjelloun et al. [47] is used. Eqs. (5) and (6), respectively, relate the capital cost of the turbine and compressor to their operating parameters, such as mass flow rate ($\dot{m}$), pressure ratio (β), isentropic efficiency η (η) and turbine inlet temperature ($T_{T,i}$). Eq. (7), which is taken from Gabbrielli and Singh [48], is used to calculate the capital cost of any type of heat exchanger. This capital cost depends on the surface area (A_{HE}) and the operating pressure (P_{HE}) of the particular heat exchanger.

$$C_T = \dot{m}_T \cdot \beta \cdot \left(\frac{392.2}{1 - \eta_T} \right) \cdot \ln(\beta) (1 + \exp(0.036 T_{T,i} - 65.66)) \quad (5)$$

$$C_C = \dot{m}_C \left(\frac{47.1}{0.92 - \eta_C} \right) \beta_C \cdot \ln(\beta_C) \quad (6)$$

$$C_{HE} = 2546.9 \cdot A_{HE}^{0.67} \cdot P_{HE}^{0.28} \quad (7)$$

The capital cost of the CO₂ pump is estimated using Eq. (8) [48] that considers the brake power ($\dot{W}_p$) and isentropic efficiency ($\eta_{i,p}$).

$$C_p = 3531.4 \cdot \dot{W}_p^{0.71} \cdot \left[1 + \left(\frac{1 - 0.8}{1 - \eta_{i,p}} \right)^3 \right] \quad (8)$$

4. Results and Discussion

4.1. Thermodynamic performance

The thermodynamic assessment of the CLC-sCO₂, considering initial design parameters presented in Table 1, revealed that this cycle has a net power output of 380.3 MW. Such output corresponded to a net efficiency of 49.3%. The detailed performance summary of the CLC-sCO₂ is shown in Table 5. As can be observed, a considerable share of total system energy input (26.4%) is utilised for the sCO₂ compression stage (compressor and pump).

Table 5. Performance summary of CLC-sCO₂

Component	Value
Thermal energy input (MW)	768.3
s-CO ₂ turbine power output (MW)	609.6
Air compression power consumption (MW)	5.4
s-CO ₂ cycle compression stage power consumption (MW)	203
CO ₂ compression for storage power consumption (MW)	23.6
Net power output (MW)	377.6
Net efficiency (%)	49.1

The thermodynamic performance of the proposed process is directly dependent on the performance of the closed sCO₂ cycle. Therefore, a parametric study of the CLC-sCO₂ was performed by varying the sCO₂ turbine inlet temperature (TIT) and turbine inlet pressure (TIP). In addition, in the sCO₂ cycle, there is an imbalance in the specific heat of the hot- and cold-side of the recuperator [49], as the specific heat of the sCO₂ stream is higher at the condition of low temperature and high pressure [50]. This difference results in a pinch point problem [26]. To compensate for this imbalance, the sCO₂ stream is split after the LTR, as shown in Figure 2b. Recompressing one stream without heat rejection [51] compensates for this imbalance, reducing the amount of waste heat in the system, and subsequently leading to a higher net efficiency. Thus, the effect of the split fraction (SF) on the net efficiency is analysed.

Figure 3 shows the variation in the CLC-sCO₂ net efficiency with split fraction. The split fraction varies from 0.25 to 0.4, indicating the fraction of total flow entering the HTR. Figure 3 reveals that the maximum efficiency is achieved when the split fraction is 0.38. This is because, at this split fraction, the approach temperature is minimised at both ends of the LTR, maximising the heat transfer rate.

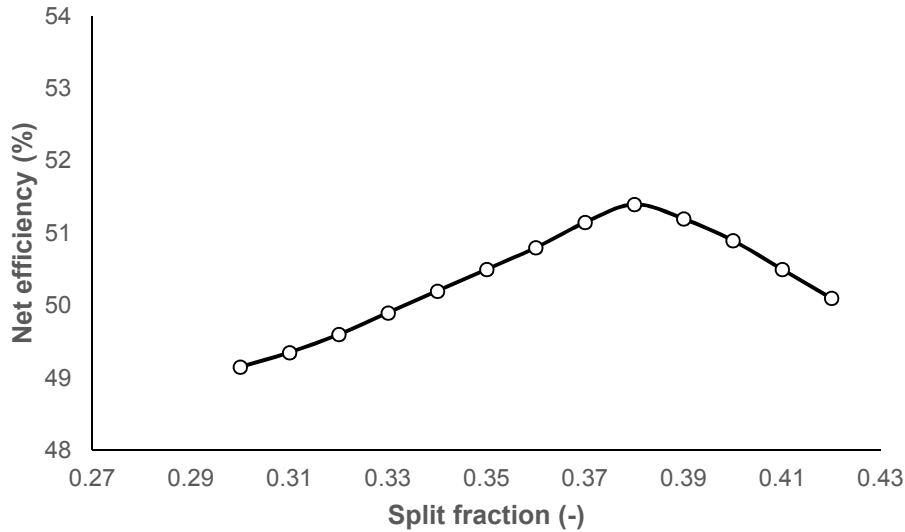
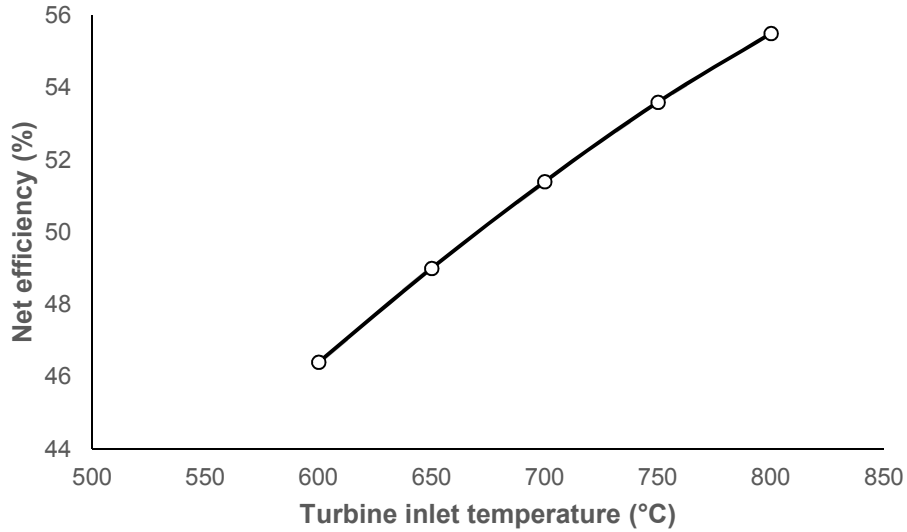
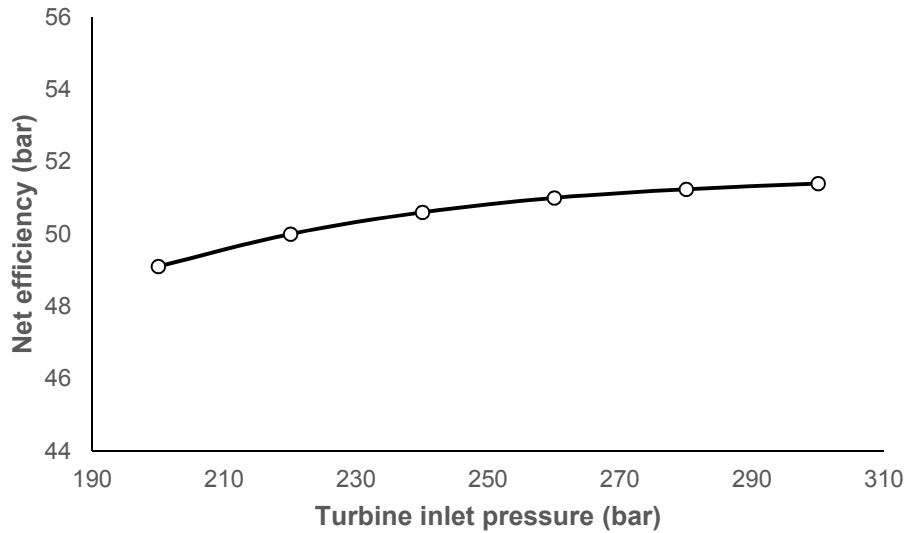


Figure 3. Effect of split fraction on the net efficiency

The effect of the sCO₂ TIT and TIP on the net efficiency of the cycle is presented in Figure 4. This parametric analysis indicated that the net efficiency of the cycle is correlated to the sCO₂ TIT and TIP. It can be seen that the correlation with the temperature is nearly linear (Figure 4a) while that with the pressure is of the second order (Figure 4b). The highest TIT of the sCO₂ cycle in this study is set at 700°C, considering the cost and lifetime of the materials under high-pressure and high-temperature conditions. Operation under such conditions was found to result in a net efficiency of 51.4%. However, further development of materials for high-temperature application would enable even higher efficiencies as further increase in TIT to 800°C resulted in a net efficiency of 55.5%. On the other hand, the analysis of the net efficiency trend in Figure 4b indicates that an increase in TIP from 200 bar to 240 bar has a more pronounced effect on the cycle performance (efficiency increases from 49.1% to 50.6%) than that from 240 bar to 300 bar (efficiency increases from 50.6% to 51.4%). It needs to be highlighted, however, that such trends for temperature and pressure, respectively, are in agreement with the results reported for sCO₂ cycles in different applications [52].



a)



b)

Figure 4. Effect of sCO₂ turbine inlet a) temperature, and b) pressure, on the net efficiency

4.2. Techno-economic performance comparison

The thermodynamic performance of the SOF-NGCC and the CLC-sCO₂ is summarised in Table 6. The same natural gas input (59470 kg/h) is used for both cycles. The analysis shows that the CLC-sCO₂ has a net power of 395 MW (Eq. 9) with a net efficiency of 51.4%. This performance is worse than that of the state-of-the-art NGCC without CO₂ capture with a net efficiency of above 62% [53]. However, it is comparable with NGCCs with CO₂ capture as PCC implementation was reported to result in an efficiency penalty of more than 8% [54]. Importantly, the net efficiencies of other configurations of CLC have been reported as below:

- CLC integrated with the humid air turbine considered by Brandvoll and Bolland (54%) [13] and by Olaleye and Wang (57.1%) [18];
- CLC-based NGCCs comprising the conventional air turbine cycle and steam cycle considered by Mantripragada and Rubin (48.9–53.2%) [10] and Naqvi and Bolland (53%) [14]; and
- CLC-sCO₂ for combined heat and power (net power efficiency of 41.3% and total efficiency of 81.7%) proposed by Chen et al. [15].

Although the sCO₂ cycle compared to the conventional power cycles, such as the air turbine cycle and steam cycle, and advanced power cycles, such as the humid air turbine cycle, has a higher thermal efficiency, the CLC-sCO₂ in this study has lower efficiency compared to the studies above. This is mainly because of different turbomachinery assumptions considered in this study. It also needs to be highlighted that, based on the turbomachinery efficiencies considered in this study, the SOF-NGCC has a net power output of 374.1 MW (Eq. 10) with an efficiency of 48.7%, which is lower by 20.9 MW compared to that of the CLC-sCO₂. This is mainly because of the power consumption of the cryogenic ASU and subsequent O₂ compression to 300 bar, which is the combustion pressure in the SOF-NGCC. In addition, compared to the required power for CO₂ compression for storage in the CLC-sCO₂, the power consumption for CO₂ purification and compression in the SOF-NGCC is much lower, 23.6 MW and 4.0 MW, respectively. This is mainly due to the higher pressure ratio for CO₂ compression (for storage) in the CLC-sCO₂ than in the SOF-NGCC, 120 and 3.4, respectively. It is worth mentioning that there is a slight difference in the CO₂ purity for storage in both cycles, 96.7% for CLC-sCO₂ and 97.6% for SOF-NGCC. This is mainly because of CPU implementation in the SOF-NGCC case. Importantly, the thermodynamic assessment of the CLC-sCO₂ considered in this study indicates that such concept could be a feasible option that would support achieving the emission reduction targets by 2050. Importantly, when natural gas is substituted with biogas, such concept will become carbon negative. Yet, the economic feasibility of such concept needs to be proven.

Table 6. Thermodynamic performance comparison between CLC-sCO₂ and SOF-NGCC

Component	CLC-sCO ₂	SOF-NGCC
Thermal energy input (MW)	768.3	768.3
High-pressure turbine power output, W_1 (MW)	-	64.4
sCO ₂ turbine power output, W_2 (MW)	658.7	576.7
Natural gas compressor power consumption, W_3 (MW)	-	3.8
Air separation unit power consumption, W_4 (MW)	-	51.9
Air/O ₂ compression power consumption, W_5 (MW)	5.4	41.4
sCO ₂ compression stage power consumption, W_6 (MW)	234.7	165.9
CO ₂ purification and compression power consumption, W_7 (MW)	-	4.0
CO ₂ compression (for storage) power consumption, W_8 (MW)	23.6	-
Gross power output, W_{gross} (MW)	424.0	475.2
Net power output, W_{net} (MW)	395.0	374.1
Net efficiency (%)	51.4	48.7

$$W_{net, CLC-sCO_2} = W_2 - W_5 - W_6 - W_8 \quad (9)$$

$$W_{net, SOF-NGCC} = W_1 + W_2 - W_3 - W_4 - W_5 - W_6 \quad (10)$$

$$W_{gross, CLC-sCO_2} = W_2 - W_6 \quad (11)$$

$$W_{gross, SOF-NGCC} = W_1 + W_2 - W_6 \quad (12)$$

The results presented in Table 5 are used as the inputs in the economic assessment of both the SOF-NGCC and CLC-sCO₂. Using these inputs, the LCOE has been calculated to assess the economic feasibility of the considered cases. The breakdown of the capital cost for both considered cases is shown in Figure 5a and Figure 5b. The capital cost of the CLC-sCO₂ is lower by £2M than that of the SOF-NGCC (£323M and £325M, respectively). The capital cost associated with the CLC reactors is almost 4 times as high as that associated with the combustors implemented in the SOF-NGCC (£41.3M and £10.8M, respectively). In addition, the higher pressure ratio of the CO₂ compressor for storage in the CLC-sCO₂ compared to the SOF-NGCC results in that compressor being more than 4 times as expensive as the corresponding unit in the SOF-NGCC (£42.2M and £9.4M, respectively). However, the cost associated with the ASU in the SOF-NGCC (£53M) results in higher capital cost of SOF-NGCC compared to CLC-sCO₂. It is worth pointing out that in both cycles, the cost associated with turbines and pumps has the highest contribution to the total capital cost.

Importantly, the specific capital cost of the CLC-sCO₂ and the SOF-NGCC are estimated to be £761/kW_{gross} and £684/kW_{gross}. Such specific capital costs are comparable with the range reported for NGCCs without CO₂ capture (£400/kW_{gross}–£700/kW_{gross}) and figures reported for NGCCs with CO₂ capture (£730/kW_{gross}–£1010/kW_{gross}) [34] . This indicates that both cycles are economically competitive to more mature technologies.

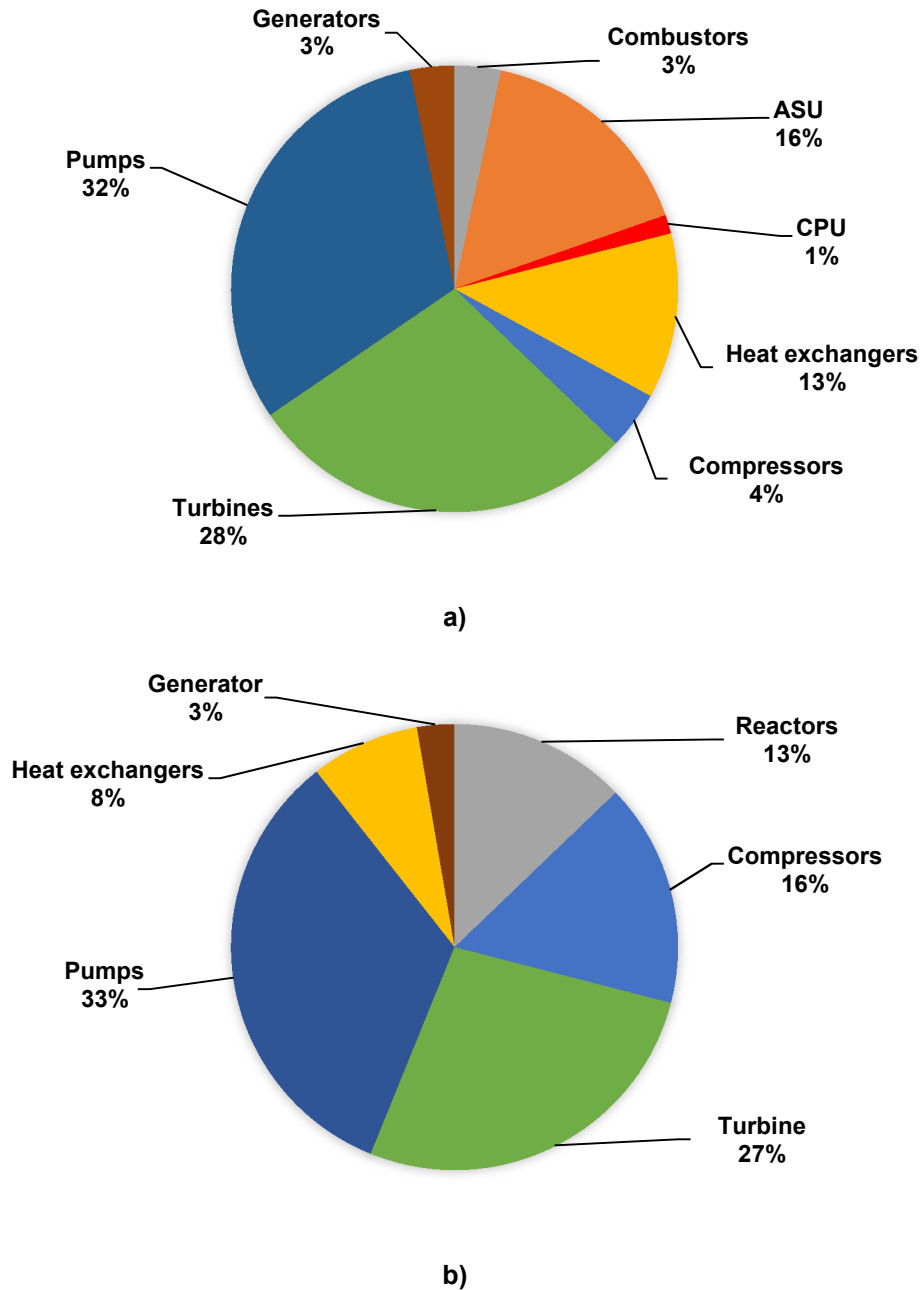


Figure 5. Breakdown of the capital cost for a) SOF-NGCC and b) CLC-sCO₂

The LCOE for the CLC-sCO₂ is £36.1/MWh, compared to £38.3/MWh for the reference SOF-NGCC cycle. This is because of the lower capital cost and higher net efficiency

of the CLC-sCO₂ compared to those of the SOF-NGCC, which results in lower fuel costs per unit of generated power. The LCOE of the CLC-sCO₂ (£36.1/MWh) is within the range reported for conventional fossil fuel power plants (£28/MWh–£55/MWh) [34]. It also needs to be highlighted that it is almost half of the cost of electricity considered for CLC with the humid air turbine (£77.5/MWh). In addition, it is superior to the LCOE of the Allam cycle (£50/MWh) [55] which is known to have the best performance among all NG-fired oxy-combustion cycles with the maximum net efficiency of 55.1% [56] and 54.6% [57] and other fossil fuel power plants with CO₂ capture (£39/MWh–£78/MWh) [34]. Therefore, further development of the CLC-sCO₂ would contribute to decarbonisation of the power sector at an affordable cost.

5. Conclusions

This study presents a novel concept of gas firing using chemical looping combustion (CLC) for oxygen supply rather than an air separation unit (ASU) in oxy-combustion systems, and the supercritical CO₂ cycle for power generation (CLC-sCO₂). A process model of the proposed system was developed in Aspen Plus®, and a parametric study was conducted by varying the inlet sCO₂ turbine conditions and split fraction to achieve the optimal performance of the CLC-sCO₂. This was followed by an economic assessment to evaluate the economic performance of the CLC-sCO₂ and staged oxy-fuel natural gas combined cycle (SOF-NGCC). It was found that the CLC-sCO₂ is characterised with a net efficiency of 51.4%, which is higher than the SOF-NGCC, which was considered as a reference, with a net efficiency of 48.7%. Lower capital cost of the CLC-sCO₂ compared to that of the SOF-NGCC (£323M and £325M, respectively), along with its lower power consumption, mainly due to no ASU, results in lower levelised cost of electricity (LCOE) by £2.2/MWh (£36.1/MWh and £38.3/MWh, respectively). Importantly, the techno-economic performance of the CLC-sCO₂ has been shown to be superior to other high-efficiency low-emission power generation cycles, such as the Allam cycle and CLC integrated with a humid air turbine, which attracts profound interest in its commercialisation.

List of Abbreviations

ASU	Air Separation Unit
PCC	Post-Combustion Capture
NGCC	Natural Gas Combined Cycle
SOF-NGCC	Staged Oxy-fuel Natural Gas Combined Cycle
sCO ₂	Supercritical CO ₂
CPU	Carbon Purification Unit
CLC	Chemical Looping Combustion
OC	Oxygen Carrier
TIT	Turbine Inlet Temperature
TIP	Turbine Inlet Pressure
LCOE	Levelised Cost of Electricity
CCS	Carbon Capture and Storage
SOFC	Solid Oxide Fuel Cell
EGR	Exhaust Gas Recycle

Nomenclature

A_{HE}	Heat exchanger surface area [m ²]
C_j	Capital cost of equipment j [£]
CF	Capacity factor [-]
FCF	Fixed charge factor [-]
FOM	Fixed operating and maintenance cost [£]
LCOE	Levelised cost of electricity [£/MWh]
LHV	Lower heating value [kJ/kg]
$\dot{m}_{fuel}$	Fuel consumption rate [kg/s]
$\dot{m}_{O_2}$	O ₂ production rate in air separation unit [kg/s]
P_{HE}	Heat exchanger operating pressure [bar]
r	Project interest rate (%)
SFC	Specific fuel cost [£/MWh]
T	Project lifetime (years)
TCR	Total capital requirement [£]

T_{Ti}	Turbine inlet temperature [°C]
VOM	Variable operating and maintenance cost [£/MWh]
$\dot{W}_{j,BRK}$	Break power output/requirement of equipment j [kW]
$\dot{W}_{net}$	Net power output [kW]
η_j	Isentropic efficiency of equipment j [%]
η_{net}	Thermal efficiency [%]
β	Pressure ratio [-]

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